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Chemistry Vocab Chapter 10. Avogadro's Number. mole. molar mass. Empirical Formula.  $6.02 \times 10^{23}$ ; counts extremely small particles. an SI base unit used to measure the quantity of matter. Mass in grams of one mole of any pure substance. Formula with smallest whole-number ratio of the elements.

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CHEMISTRY: Chapter 10 Prep-Test Matching Match each item with the correct statement below. a.

calorimeter d. temperature b. calorie e. specific heat c.

joule f. heat \_\_\_\_ 1. quantity of heat needed to raise the

temperature of 1 g of water by 1 C \_\_\_\_ 2. SI unit of

energy \_\_\_\_ 3. energy transferred between 2 objects

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