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Bonding in Polyatomic Ions and

Compounds How To Write Ionic Formulas With Polyatomic Ions

How to Name Ionic Compounds with Polyatomic Ions

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Lewis Structures for Polyatomic Ions ~~Chemistry Lesson:~~

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S2E5 - Naming Ionic Compounds with Polyatomic Ions; List of Polyatomic Ions to Know **Naming Ionic and Molecular**

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~~and writing formulas for polyatomic ionic compounds~~ How To:  
Polyatomic Ions ~~Notes: Ionic Compounds with Polyatomic~~  
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4.1 Polyatomic ions (SL) *Ionic bonding and ions : from*  
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7. Write formulas for the following ionic compounds. Note that each includes a polyatomic ion. a) lithium phosphate b) ammonium oxide c) barium hydroxide  $\text{Li}_3\text{PO}_4$   $(\text{NH}_4)_2\text{O}$   $\text{Ba}(\text{OH})_2$  d) calcium cyanide e) sodium chlorate f) potassium sulfate  $\text{Ca}(\text{CN})_2$   $\text{NaClO}_3$   $\text{K}_2\text{SO}_4$  8. In question 3, you were asked the difference between calcium nitride and ...

*Information : Polyatomic Ions*

ChemQuest 20 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_

Information: Polyatomic Ions The word, "polyatomic" means

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“many atoms”. A polyatomic ion, therefore, is an ion that is made of more than one atom. An example of a polyatomic ion is the sulfate ion,  $\text{SO}_4^{2-}$ . Sulfate is composed of one sulfur

### *Information: Polyatomic Ions*

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