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Group 1 contains metals which lose electrons, while Group 17 members are nonmetals and gain electrons. Na^+ has a higher ionization energy as it has 11 protons holding 10 electrons while Na has 11 protons holding 11 electrons.

ANSWERS TO TOPIC 5
Review Questions 12. 21.
24. 27. 30. 33. 36. 39. 42.

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45. 48. 51. 54. 57. 60. 63.
66. 69. 72. 4 3 3 2 ...

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Answers equilibrium.

Colorless N_2O_4 reacts to form brown NO_2 . As the reaction proceeds toward equilibrium, the color of the mixture darkens due to the increasing concentration of NO_2 . The formation of NO_2 from N_2O_4 is an endothermic reaction. 13.1 Chemical Equilibria – Chemistry fizzy

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Metal marble crossword

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new installment in the
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Answers

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you tell that the reactions
are producing a gas?

2a What is the gas that the
reactions are producing?

b Describe how they
could test for this gas.

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3a Which metal reacted the fastest? b How did you work out your answer to part a? 4

Danny made a mistake in the experiment.

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