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1. Study the flow chart below and answer the questions that follow: 4. Name reagent

Z. 5. Describe the process which takes place in step 2. 6. Identify the white solid.

2. a) Starting from solid magnesium oxide, describe how a solid sample of magnesium hydroxide can be prepared. b) Give one use of magnesium hydroxide. 3.

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Sent! We will get in touch soon. If Urgent,
call or text 0716 858334 / 0712 054195.

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FORM 2 CHEMISTRY TIME: 2 Hrs

INSTRUCTIONS Attempt all the question
in the question in the spaces provided All
the working must be clearly shown where
appropriate. Negligence will be penalized.

QUESTIONS 1.The table below shows
some information concerning partials of
certain elements. Element Atomic number
Number of electrons Mass Number P 19
18 39

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Chemistry. Form 2. The Structure of the Atom and the Periodic Table. 1.1 The structure of the atom. When scientists started exploring matter, they realised that matter can be divided into smaller and still smaller particles. They called the smallest particle an 'atom'. The name 'atom' was derived from the Greek word 'atomos', meaning 'indivisible'.

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INSTRUCTIONS TO CANDIDATES 1.

Answer all the questions in the spaces provided. 2. Mathematical tables and electronic calculators may be used. 3. All

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Answers must be clearly shown where necessary 1. The table below shows the pH values of solutions J to N.

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1. What is the first element on the periodic table? 2. What is the centre of an atom called? 3. True or false? Acids have a pH level below 7. 4. What is the main gas found in the air we breathe?

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Answers (8-10) in Chemistry Lab. Charles Thatcher, Getty Images If there had been a million dollars on the line for knowing all the answers, you would have just missed the prize. It's OK to miss a question or two. That just shows you have other interests! You have a good command of chemistry.

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Answers? Answer: Violet. 2) The modern atomic mass unit is based on ____.

Answer: Carbon-12.

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