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Chapter 2. 1. A solution is a
homogeneous mixture of two or 9.
more chemically non-reacting

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Substances. The components of a solution generally cannot be separated by filtration, settling or centrifuging. 2. A solution may be classified as solid, liquid or a gaseous solution. 3.

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component which is small in quantity is called solute. If both components are in same physical state.

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The difference in boiling points of solution and pure solvent is called elevation in boiling point . Solutions: Solutions are the homogeneous mixtures of two or more than two components. Binary solution: A solution having two components is called a binary solution.

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